



AVIRAL CLASSES
CREATING SCHOLARS

JEE (ADVANCED), PMT & FOUNDATIONS

UTS- NEET -2020

MOCK TEST-03 SOLUTION

ANSWER KEY

Que.	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20
Ans.	3	2	3	2	2	2	2	3	3	4	3	3	3	2	4	4	3	1	3	4
Que.	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40
Ans.	3	3	2	3	3	3	2	4	3	4	4	2	1	1	1	3	3	2	4	3
Que.	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60
Ans.	2	4	2	3	2	3	3	2	2	3	4	3	2	2	4	1	3	4	2	1
Que.	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80
Ans.	3	3	4	1	4	1	2	4	3	2	4	1	4	2	3	3	1	1	4	4
Que.	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100
Ans.	4	3	3	2	3	4	3	1	3	4	4	4	4	3	2	3	3	3	3	2
Que.	101	102	103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118	119	120
Ans.	2	3	2	1	4	2	4	2	2	2	3	3	4	2	3	4	4	4	2	1
Que.	121	122	123	124	125	126	127	128	129	130	131	132	133	134	135	136	137	138	139	140
Ans.	4	1	2	1	3	1	2	4	1	1	2	1	4	3	2	4	3	3	3	2
Que.	141	142	143	144	145	146	147	148	149	150	151	152	153	154	155	156	157	158	159	160
Ans.	2	2	3	4	1	4	3	3	4	4	4	3	2	3	1	2	3	4	2	3
Que.	161	162	163	164	165	166	167	168	169	170	171	172	173	174	175	176	177	178	179	180
Ans.	2	1	4	3	1	4	4	3	2	2	3	1	1	1	2	3	2	3	3	1

1.

Component of \vec{x} on $\vec{y} = x \cos \theta$

$$= \frac{\vec{x} \cdot \vec{y}}{y} = \frac{(\vec{a} - \vec{b}) \cdot (\vec{a} + \vec{b})}{|\vec{a} + \vec{b}|}$$

$$= \frac{a^2 - b^2}{\sqrt{a^2 + b^2}}$$

2.

$$KE = \frac{J^2}{2I}$$

3. (3)

4. (2)

5.

$$x = 5.0 \cos(2\pi t + \pi) = -5.0 \cos 2\pi t$$

$$\text{Velocity} = \frac{dx}{dt} = 10\pi \sin 2\pi t$$

$$\text{Acceleration} = \frac{d^2x}{dt^2} = 20\pi^2 \cos 2\pi t$$

\therefore At $t = 1.5$ sec,

$$x = -5.0 \cos 3\pi = -5.0 (-1) = 5$$

$$\text{Velocity} = \frac{dx}{dt} = 10\pi(\text{zero}) = 0$$

$$\text{Acceleration} = \frac{d^2x}{dt^2} = 20\pi^2(-1) = -20\pi^2$$

6.

According to work energy theorem

$$W_{mg} + W_F = \Delta K$$

$$Mg(h + S) - FS = 0 \quad [K_i = K_f = 0]$$

$$FS = Mg(h + S)$$

$$F = Mg \left(1 + \frac{h}{S} \right)$$

7.

$$\vec{v}_B = 2\vec{v}_A$$

$$\vec{v}_{CA} = x\hat{i} \Rightarrow \vec{v}_C - \vec{v}_A = x\hat{i} \quad \dots (1)$$

$$\vec{v}_{CB} = x\hat{j} \Rightarrow \vec{v}_C - \vec{v}_B = x\hat{j}$$

$$\Rightarrow \vec{v}_C - 2\vec{v}_A = x\hat{j} \quad \dots(2)$$

Multiply eq. (1) by (2) and subtract eq. (2) from it

$$2\vec{v}_C - 2\vec{v}_A - (\vec{v}_C - 2\vec{v}_A) = 2x\hat{i} - x\hat{j}$$

$$\vec{v}_C = 2x\hat{i} - x\hat{j}$$

$$\text{dir}^n = E \tan^{-1} \left(\frac{x}{2x} \right) S = E \tan^{-1} \left(\frac{1}{2} \right) S$$

8.

$$u = -25 \quad v = -50$$

$$\frac{1}{f} = \frac{1}{v} - \frac{1}{u} = \frac{1}{50}$$

$$\therefore P = 2D$$

9. (3)

10.

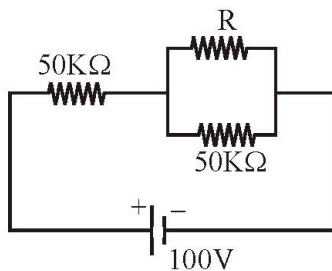
No conduction is found between P & Q so both P & Q will be n-type or p-type. Therefore R will be base.

11.

$$a_{\text{after } 50 \text{ s}} = \frac{F}{M_{\text{after } 50 \text{ s}}} = \frac{v \frac{dm}{dt}}{1500 - 10 \times 50}$$

$$= \frac{5000 \times 10}{1000} = 50 \text{ m/s}^2$$

12.



$$V_2 = \frac{R_2}{R_1 + R_2} \times V$$

$$\frac{100}{3} = \frac{50 \times R}{50 + R + 50} \times 100$$

$$\Rightarrow R = 50 \text{ K}\Omega$$

13.

$$\Delta x = \frac{5\lambda}{2}$$

$$\frac{\Delta x}{\lambda} = \frac{\Delta \phi}{2\pi}$$

$$\Delta \phi = 5\pi$$

14.

$$T_{1/2} = 0.7 RC$$

$$R = \frac{T_{1/2}}{0.7C} = \frac{70 \times 10^{-3}}{0.7 \times 2 \times 10^{-6}} = \frac{10^5}{2} = 5 \times 10^4$$

$$R = 50 \times 10^3 \Omega = 50 \text{ k}\Omega$$

15.

$$\frac{5V}{4L} = \frac{2V}{2L} + 100$$

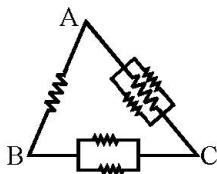
$$\frac{V}{2L} \left(\frac{5}{2} - 2 \right) = 100$$

$$\frac{V}{2L} = 200$$

16. (4)
17.

$$F_{avg} = \frac{\Delta p}{\Delta t} = \frac{2M(u+v)}{t}$$

18.



$$R_{AB} = r, \quad R_{BC} = \frac{r}{2}$$

$$R_{AC} = \frac{r}{3}$$

19.

$$R_F \propto \frac{A}{MS} \propto \frac{1}{\rho S} \quad [T, T_0, e_r, r = \text{same}]$$

20.

$$\vec{F}_{net} = 0$$

$$\vec{F} + \vec{F}_{remaining} = 0$$

$$\vec{F}_{remaining} = -\vec{F}$$

21.

$$T_P = M_{below} \times (g + a) = (3 + 5)(10 + 10) = 160N$$

22. (3)
23.

$$a_r = \frac{g \sin \theta}{1 + K^2/r^2}$$

For shell $\frac{K^2}{r^2} = \frac{2}{3}$

$$a_r = \frac{3}{5} g \sin \theta \quad \text{----- (1)}$$

but $a_s = g \sin \theta \quad \text{----- (2)}$

dividing the two equations

$$\frac{a_r}{a_s} = \frac{3}{5}$$

24. (3)
25.

For isothermal process

$$P \propto \frac{1}{V}$$

Volume expands to double so pressure will be halved for adiabatic process

$$PV^\gamma = \text{const}$$

$$\frac{P_1}{P_2} = \left(\frac{V_2}{V_1}\right)^{\frac{5}{3}}$$

$$\frac{P/2}{P_2} = \left(\frac{16V}{2V}\right)^{\frac{5}{3}}$$

$$P_2 = \frac{P}{64}$$

26.

$$8 = I_B R_B + V_{BE}$$

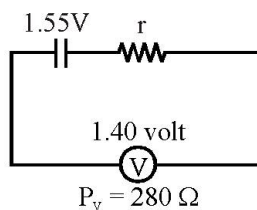
$$\Rightarrow 8 = \left(\frac{4 \times 10^{-3}}{100}\right) R_B + 0.6$$

$$\Rightarrow R_B = 185 \text{ k}\Omega$$

27.

Thermal Tension $F = YA \alpha \Delta \theta$

28.



$$1.40 = 1.55 - ir \quad \text{...(i)}$$

$$\text{but } i = \frac{1.55}{280 + r} \quad \text{...(ii)}$$

On solving we get
 $r = 30 \Omega$

29.

$$v_{\text{rms}} \propto \frac{1}{\sqrt{M_{\text{W}}}} \Rightarrow \frac{v_1}{v_2} = \sqrt{\frac{M_{\text{W}_2}}{M_{\text{W}_1}}}$$

$$\Rightarrow \frac{v}{\sqrt{2v}} = \sqrt{\frac{M_{\text{W}_2}}{32}} \Rightarrow M_{\text{W}_2} = 16(\text{CH}_4)$$

30.

$$\frac{dQ}{dt} = \frac{KA\Delta T}{\ell} \quad (\because A \propto r^2)$$

$$\frac{dQ}{dt} \propto \frac{r^2}{\ell}$$

31. (4)

32.

$$\frac{\phi_1}{\phi_2} = \frac{\lambda_2}{\lambda_1} = \frac{2}{1}$$

33.

Total distance covered by two wheels is same

$$2\pi r_1 n_1 = 2\pi r_2 n_2$$

$$\text{or } 0.5 \times 200 = 0.1 n_2 \Rightarrow n_2 = 1000 \text{ rpm}$$

34.

$$\text{Phase diff.} = \frac{\pi}{3} - \frac{\pi}{6} = \frac{\pi}{6}$$

$$Z = \frac{200}{2} = 100 \Omega \quad \sin \phi = \frac{X}{Z}$$

$$X = \sin\left(\frac{\pi}{6}\right) \times 100 = 50 \Omega$$

I leads E so reactance is capacitive.

35.

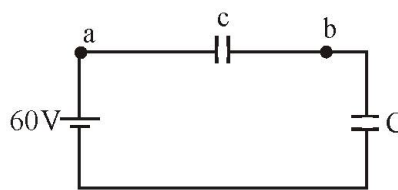
Volume remains same $\Rightarrow AL = A'(2L)$

$$A' = \frac{A}{2}; R_i = \frac{\rho L}{A}$$

$$R_f = \frac{\rho(2L)}{(A/2)} = 4R_i$$

$$\% \text{change} = \frac{4R_i - R_i}{R_i} \times 100 = 300\%$$

36.



$$V_{ab} = \frac{60}{2} = 30V$$

37.

$$\text{Since } P' = n^3 P$$

$$\therefore n = 3 \quad P' = 27P$$

38.

Apply energy conservation

Energy at surface = Energy at height

$$\frac{-GM_e m}{R_e} + \frac{1}{2} m \left(\frac{v_e}{2} \right)^2 = \frac{-GM_e m}{(R_e + h)}$$

$$\text{but } v_e^2 = \frac{2GM_e}{R_e}$$

$$\text{Hence } \frac{-GM_e m}{R_e} + \frac{1}{4} \frac{GM_e m}{R_e} = \frac{-GM_e m}{R_e + h}$$

$$\text{Hence } R_e + h = \frac{4}{3} R_e$$

$$\boxed{h = \frac{R_e}{3}}$$

39.

$$n' = n \frac{V + V_c}{V - V_c}$$

$$n' = 700 \left(\frac{352 + 2}{352 - 2} \right)$$

$$n' = 708 \text{ Hz}$$

40.

near A \rightarrow +ve potentialnear B \rightarrow -ve potential

$$a_A = (+)$$

$$a_B = (-)$$

Also potential at both $-\infty$ & $+\infty$ are +ve $\Rightarrow a_A$ dominates a_B So $|a_A| > |a_B|$

41.

$$\frac{2u_y}{g} = \frac{2u_x u_y}{g} = \frac{\sqrt{3}}{g}$$

$$\therefore u_x = 1 \text{ \& } u_y = \frac{\sqrt{3}}{2}$$

$$\therefore \tan\theta = \frac{u_y}{u_x} = \frac{\sqrt{3}}{2} \Rightarrow \theta = \tan^{-1}\left(\frac{\sqrt{3}}{2}\right)$$

42.

$$M = \frac{f_o}{f_e} = 5 \quad \therefore f_o = 5f_e$$

$$f_o + f_e = 24 \quad 6f_e = 24$$

$$\boxed{f_e = 4} \quad \text{and} \quad \boxed{f_o = 20}$$

43.

$$\text{sol. } PV^\gamma = \text{const.} \Rightarrow v^\gamma \propto \frac{1}{P}$$

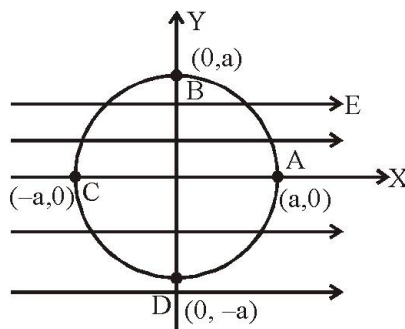
$$TV^{\gamma-1} = \text{constant} \Rightarrow V^{\gamma-1} \propto \frac{1}{T}$$

44.

$$e = -L \frac{di}{dt}$$

$$= -6 \times 10^{-3} \left(-\frac{(7-5)}{(5-2) \times 10^{-3}} \right) = 4V$$

45.



$$V_B = V_D$$

in direction of electric field potential will be decreased

$$\text{So } V_C > V_B = V_D > V_A$$

46. (3)

47. (3)

48.

2-amino pyridine is not present in nucleic acid

49. (2)

50.

$$M_1 V_1 = M_2 V_2$$

$$V_2 = \frac{M_1 V_1}{M_2} = \frac{0.25 \times 40}{0.1} = 100 \text{ cc}$$

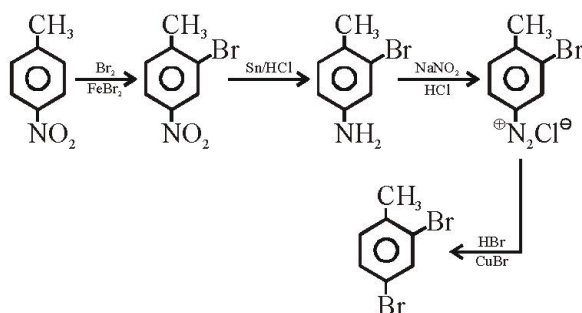
$$\therefore V_2 - V_1 = 100 - 40 = 60 \text{ cc}$$

51. (4)

52. (3)

53. (2)

54.



55.

$$\text{H}_3\text{PO}_4 \quad n = 2 \quad \frac{48}{2} = 24$$

$$\text{NaOH} \quad n = 1 \quad \frac{40}{1} = 40$$

56. (1)
57. (3)
58. (4)
59.

$$V_1^H = 2.18 \times 10^6 \times \frac{z}{n} = 2.18 \times 10^6 \text{ m/s}$$

$$\Delta V = 2.18 \times 10^6 \times \frac{0.1}{100} = 2.18 \times 10^3 \text{ m/s}$$

$$\Delta x \times m \times \Delta V = \frac{h}{4\pi}$$

$$\Delta x \times 9.1 \times 10^{-31} \times 2.18 \times 10^3 = \frac{6.62 \times 10^{-34}}{4 \times 3.14}$$

$$\Delta x = 2.66 \times 10^{-8} \text{ m.}$$

60.

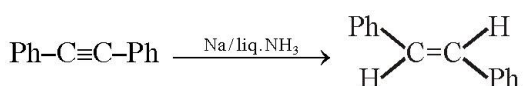
$$\alpha = \frac{\wedge_M}{\wedge_M^o}$$

$$\alpha = \sqrt{\frac{K_a}{C}} = \sqrt{\frac{1.75 \times 10^{-5}}{0.01}} = \sqrt{1.75 \times 10^{-3}} = 4.2 \times 10^{-2}$$

$$\wedge_M = \alpha \wedge_M^o = 4.2 \times 10^{-2} \times 390.6 \times 10^{-4}$$

$$K = \frac{\wedge_M \times M}{1000} = \frac{4.2 \times 10^{-2} \times 390.6 \times 10^{-4} \times 0.01}{1000}$$

61. (3)
62. (3)
63.



(Birch Reduction)

64.

$$K_1 = \frac{[\text{CO}]^2 [\text{H}_2\text{O}]^2}{[\text{CO}_2]^2 [\text{H}_2]^2}$$

$$K_2 = \frac{[\text{CH}_4][\text{H}_2\text{O}]}{[\text{CO}][\text{H}_2]^3}$$

$$K_3 = \frac{[\text{CO}_2][\text{H}_2]^4}{[\text{CH}_4][\text{H}_2\text{O}]^2}$$

$$K_3 = \frac{1}{K_2 \times K_1^{1/2}}$$

hcp and ccp both have same C.N. = 12.

66. (1)
67. (2)
68.

Chlorobenzene on hydrolysis gives phenol.

69.

$$K_h = \frac{K_w}{K_a}$$

$$K_a = \frac{K_w}{K_h} = \frac{10^{-14}}{3 \times 10^{-7}}$$

70.

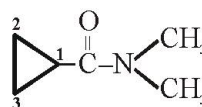
$$\Delta H = E_{af} - E_{ab}$$

$$0 = E_{af} - E_{ab}$$

$$E_{af} = E_{ab}$$

71. (4)

72.

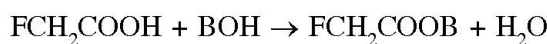


N,N-Dimethyl Cyclopropane Carboxamide

73.

Reactivity of halide \propto stability of carbocation

74.



10m mol 5m mol 0 0

5m mol 0m mol 5m mol

$$\text{For acidic buffer } \text{pH} = \text{PK}_a + \log \frac{[\text{Salt}]}{[\text{Acid}]}$$

$$\text{PK}_a = -\log K_a$$

75.

$$\Pi = \text{CST}$$

$$= 1.0 \times 10^{-3} \times 0.0821 \times 298 \text{ atm.}$$

$$= 18.6 \text{ mm}$$

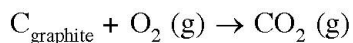
76. (3)

77. (1)

78. (1)

79.

65.

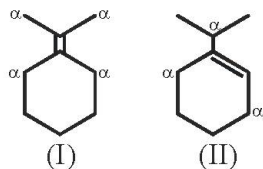


It represent enthalpy of combustion of graphite as well as enthalpy of formation of $\text{CO}_2(\text{g})$

80. (4)

81. (4)

82.

 $\alpha\text{-H} \Rightarrow 10$ $\alpha\text{-H} \Rightarrow 5$

$$\Rightarrow 10 + 5 = 15 \alpha\text{-Hydrogen}$$

83. (3)

84.

In HIO_4 and H_3BO_3 central atom present in highest oxidation state so act as oxidant only.

85. (3)

86. (4)

87.

Q and R shows aromatic character due to Huckle rule.

88. (1)

89. (3)

90. (4)